Synthesis of 4‑Membered Carbasugars by Way of Stereoselective SmI2‑Mediated Aldehyde−Alkene Cyclization

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S Supporting Information

Carbohydrates mimetics are structurally altered analogues
of carbohydrate designed to simulate the shape and most functionalities of the natural substrates in the ground state or in the transition state with the goal of discovering drug candidates or useful biological probes.¹ The most common structural modification performed is the replacement of the endocyclic and/or the glycosidic oxyge[n](#page-5-0) atom by a heteroatom or by a carbon atom. Carbasugars, named "pseudosugars" in the pioneering article of McCasland et al., 2 are one of the most important class of glycomimetics. 3 In these analogues of furanoses or pyranoses, the ring ox[yg](#page-5-0)en is replaced by a methylene group (branched-chain [cy](#page-5-0)clitols). Carbasugars are attractive in the context of drug discovery because of their stability toward endogenous degradative enzymes as well as their interesting biological properties mainly as antibiotics and as glycosidase inhibitors.^{3,4} For example, acarbose $(Glucobay)$ ⁵ and voglibose $(Basen)^6$ are now clinically useful therapeutic agents to control diabet[es.](#page-5-0) If not surprisingly most carbocycli[c](#page-5-0) sugar mimetics are [ca](#page-5-0)rbafuranoses or carbapyranoses, few examples of seven- $,7$ eight- $,8$ and even nine-membered⁹ carbasugars have been reported recently in the literature. Beyond the synthetic [c](#page-5-0)halleng[es](#page-5-0), the main motivation for th[e](#page-5-0) synthesis of such medium-ring carbasugars is the access to a diversity of conformations other than the traditional chair boat of six-membered rings. The original distributions of hydroxyl groups thus obtained in addition to the fine-tuning of the hydrophobic−hydrophilic balance are thought to be of likely significance for receptor recognition purposes. In contrast, to our knowledge, no example of 4-membered carbasugars has been reported in the literature to date. These compounds are attractive as they offer opposite and complementary structural features compared to the corresponding medium-ring ana-

logues including conformational rigidity and molecular simplicity. On the basis of these considerations and in conjunction with our continuing studies on original glycomimetics,¹⁰ we have synthesized the first members of a new class of 4-membered carbasugars, "carbaoxetanoses" from vitamin C $(1).$ ¹¹ [O](#page-5-0)ur retrosynthetic analysis takes advantage of the chirality of 1, which provides two stereogenic centers (C1 and C2[\) o](#page-6-0)f the final compounds and secures the stereocontrol of the key reductive coupling reaction from γ , δ -unsaturated aldehydes III (Scheme 1). This pivotal step based on an intramolecular SmI₂-mediated carbonyl–alkene coupling¹² was expected to provide cyclobutanols II with the four contiguous stereogenic centers of the target molecules of type [I](#page-6-0) with predictable stereocontrol. The absolute configuration of the resulting alcohol at C3 may be directly inverted to provide

Scheme 1. Retrosynthetic Analysis

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access to further members of carbaoxetanoses of type I′. 4- Membered carbasugars of type I and I′ may be seen as simplified structural analogues of hexopyranoses in the D-gluco and D-galacto series, respectively lacking the endocyclic oxygen and the anomeric carbon. They may be seen also as analogues of pentofuranoses in the D-arabino (type I) or D-lyxo (type I') series lacking the endocyclic oxygen (Scheme 2).

Scheme 2. α -D-Glucopyranose (2), α -D-Arabinofuranose (3), and Their Corresponding Carbasugar Analogues 5a-Carba- α -D-glucopyranose $(4)^3$ and Carbaoxetanose 5 (this study)

The synthesis began with the O-benzylation of γ -lactone 6 obtained in three steps from vitamin C according to a procedure reported by Eschenmoser et al. (Scheme 3). 13

^aReagents and conditions: (a) BnBr (5 equiv), CaSO₄ (5 equiv), Ag₂O (4 equiv), CH3CN, rt, 48 h, 79%; (b) DIBAL-H (1.9 equiv), THF, −78 °C, 2.5 h, 85%; (c) 11 (1.2 equiv), PhCO2H (0.03 equiv), CH₂Cl₂, reflux, 97% $(Z/E, 1:2)$; (d) $(CICO)$ ₂ (2.2 equiv), DMSO (4.6 equiv), NEt₃ (5.5 equiv), CH₂Cl₂, -78 to -20 °C, 79% [(Z/E)-10)], 79% [(Z)-10)], 74% [(E)-10)].

Reduction of 2,3-di-O-benzyl-L-threonolactone $(7)^{14}$ using DIBAL-H provided the corresponding lactol 8, which was readily converted to the α , β -unsaturated esters 9 obta[ine](#page-6-0)d as a separable 1:2 mixture of Z- and E-isomers by treatment with ethyl(triphenylphosphoranylidene)acetate (11) in the presence of catalytic amount of benzoic acid.¹⁵

Swern oxidation performed on the mixture of alcohols 9, or on the corresponding pure (E) - or (Z) (Z) (Z) -stereoisomers, afforded the desired γ,δ-unsaturated aldehydes 10 in good yields. The key SmI2-mediated 4-exo-trig radical cyclization was first evaluated from the pure alkene (E) -10. In a first attempt, treatment of (E) -10 with SmI₂ at −78 °C in the presence of HMPA did not lead to the formation of the expected cyclobutanol ring (Table 1, entry 1).¹⁶ The only identified product was the conjugated aldehyde 13 resulting from the

Table 1. $SmI₂$ -Mediated 4-exo-trig Radical Cyclization of 10^a

	റ BnO 10	CO ₂ Et Sml ₂ additives THF, -78°C OBn	HO, CO ₂ Et 2 ОВn BnO 12 de > 96%	OBn 13	CO ₂ Et
entry	10 (Z/E)	SmI ₂ (equiv)	additives (equiv)	time (min)	12^b (yield %)
$\mathbf{1}$	0/1	3	HMPA (6.8)	120	
2^c	0/1	3	HMPA (6.8)	960	15
3^d	0/1	3	MeOH(3)	960	14
$\overline{4}$	0/1	3	MeOH(3)	150	28
5	0/1	3	MeOH(22)	135	21
6	0/1	3	MeOH(3)	10	42
7	1/0	3	MeOH(3)	10	35
8	1/2	$\overline{4}$	H ₂ O/LiBr (8/12)	5	48
9	1/2	$\overline{4}$	HFIP/LiBr (8/12)	5	67
10 ^e	1/2	4	HFIP/LiBr (8/12)	19^e	57

^aReaction performed with 0.3 mmol of 10. ^bIsolated yield. ^c-78 °C to rt. ${}^{d}0$ °C to rt. e Reaction performed with 1.2 mmol of 10.

elimination of a benzyloxy group. The formation of this sideproduct, which is favored by conjugation, further highlights that γ , δ -unsaturated aldehyde 10 is a challenging substrate for 4-exotrig radical cyclization reactions. Small amount of the desired cyclized product 12 (∼15%) could nevertheless be obtained by increasing the reaction time and temperature (entry 2) or using $MeOH¹⁷$ as a proton source (entry 3). Cyclization of 10 was found to proceed with complete anti selectivity,^{12a} affording the desired [c](#page-6-0)yclobutanol 12 as a single diastereoisomer with the desired stereochemistry. The absolute configur[atio](#page-6-0)n of the two new stereogenic centers were unambiguously determined by 2D COSY and NOESY NMR experiments. In particular, the definite NOE effects between H-1 and H-3, and between H-2 and H-4 were crucial to establish the R-configuration at C-3 and C-4 (see Supporting Information). In the presence of MeOH, the yield of the cyclization process could be tripled to 42% by shorte[ning the reaction time to](#page-5-0) 10 min (entries 3−6). Addition of more equivalents of MeOH (entry 5) did not improve the efficiency of the coupling reaction. It is noteworthy that the coupling process performed from the TBDMS protected analogues of 10 led only to partial recovery of starting material. We then evaluated the impact of the double bond configuration on the cyclization outcome. Pleasingly, treatment of alkene (Z) -10 under our first optimized reaction conditions led to the desired cyclobutanol 12 in high diastereoselectivity and in similar yield than from the corresponding E-stereoisomer (entries $6-7$). The SmI₂mediated radical coupling reaction was then performed directly from the E/Z mixture of 10, leading to a more efficient synthetic sequence and better overall yields, by avoiding the separation of diastereoisomers. Further optimization was performed with a combination of $SmI₂$ and $LiBr/proton$ source additives¹⁸ by analogy with the reaction conditions reported recently for the cross-coupling of nitrones with β -silyl acrylates (entries $8-10$ $8-10$).¹⁹ Similar to the results obtained by Py et al.¹⁹ for this cross-coupling, the optimum reaction conditions were obtained with [the](#page-6-0) use of hexafluoroisopropanol (HFIP) as [a](#page-6-0) proton source, with 12 being isolated in 57−67% yield (entries

9 and 10). The optimum reaction time was found to be related to the amount of substrate used with one minute for ca. 0.06 mmol of 10 as a rule of thumb (entries 9 and 10).

The complete anti selectivity observed for the 4-exo-trig cyclization of aldehyde 10 can be rationalized considering the four possible transition state structures depicted in Scheme 4.^{12a} From E- or Z-alkene substrates 10, syn- (E) -TS or syn- (Z) -

S[che](#page-6-0)me 4. Possible Transition States for the Intramolecular SmI2-Mediated Aldehyde−Alkene Coupling

TS leading to the 1,4-cis product are disfavored mainly by electrostatic repulsion between the O1 oxygen lone pairs and the developing electron density in α to the ester group. The all anti- (E) -TS and all anti- (Z) -TS leading to the all-trans product 12 are electronically and sterically favored since the abovementioned electrostatic interactions as well as the steric hindrance between vicinal substituents are minimized.

Having in hand the key intermediate 12, we first prepared 4 membered carbasugar 15 with a hydroxyethyl group at C4 as its synthesis required only two steps. Reduction of ester 12 with LAH followed by catalytic hydrogenolysis of the benzyl protecting groups provided tetrol 15 (Scheme 5). Synthesis of 5 that may be seen as a structurally simplified analogue of Dglucopyranose or α -D-arabinofuranose required the dehomologation of the side chain in 14 by one methylene unit. As a prelude to the dihydroxylation−dehomologation sequence, alkene 16 was synthesized from alcohol 14 via Grieco elimination.²⁰ Treatment of alcohol 14 with o -nitrophenylselenocyanate and tributylphosphine provided a selenide derivative. [Th](#page-6-0)is intermediate was directly oxidized with Davis oxaziridine^{20b} to give the corresponding selenoxide in which a Cope-type elimination took place to afford the desired to afford the desire[d al](#page-6-0)kene 16.

Protection of the secondary alcohol as a TBS ether to provide compound 17 in 77% yield from 14 was found to be necessary for effective olefin cleavage. Olefin 17 was treated under Lemieux−Johnson oxidative cleavage conditions to give the primary alcohol 18 in 82% yield after reduction with NaBH4. The deprotected carbaoxetanose 5 was finally obtained in quantitative yield after removal of the TBS group in 18 using TBAF followed by hydrogenolysis of the benzyl protecting groups. Access to other members of carbaoxetanoses as simplified mimetics of D-galactopyranose or D-lyxofuranose was easily achieved by inversion of configuration at C3 (Scheme 6). Benzylation of the primary hydroxyl group in 18 followed by treatment of the resulting product 20 with TBAF

Scheme 5^a

^aReagents and conditions: (a) SmI_2 (4 equiv), LiBr (12 equiv), HFIP (8 equiv), THF, −78 °C, 5 min, 67%; (b) LAH (1.5 equiv), THF, rt, 2.5 h, 68%; (c) Pd/C 10%, HCO₂H, H₂, EtOH, rt, 16 h, quant.; (d) o-NO2PhSeCN (2.2 equiv), n-Bu3P (2.2 equiv), THF, 35 min; (e) Davis oxaziridine (1.2 equiv), CH_2Cl_2 , 0 °C, 40 min, 84% in two steps; (f) TBSCl (1.5 equiv), DMAP (0.5 equiv), NEt₃ (2 equiv), CH₂Cl₂, rt, 14 h, 77%; (g) OsO₄ (0.04 equiv), NaIO₄ (1.9 equiv), THF/H₂O (4:1), rt, 5 h; (h) NaBH₄ (1.2 equiv), MeOH, rt, 15 min, 82% in two steps; (i) TBAF (5 equiv), THF, rt, 4 h, quant.; (j) Pd/C 10%, HCO₂H, H₂, EtOH, rt, 16 h, quant.

Scheme 6^a

a Reagents and conditions: (a) NaH (1.3 equiv), BnBr (1.2 equiv), THF, rt, 18 h, 49% (73% based on recovered starting material); (b) TBAF (5 equiv), THF, rt, 3 h, 97%; (c) DMP (1.6 equiv), CH_2Cl_2 , rt, 92%; (d) L-selectride (1.1 equiv), THF, −78 °C, 1 h, 67%; (e) Pd/C 10%, HCO₂H, H₂, EtOH, rt, 20 h, quant.

afforded the key alcohol intermediate 21. Oxidation of 21 with Dess−Martin periodinane reagent provided the corresponding ketone 22, which was reduced with L-selectride. This diastereoselective process gave the desired alcohol 23 in 62% yield from the corresponding C3 epimer 21. Deprotection of 23 by hydrogenolysis under acidic conditions afforded cyclobutanic pseudogalactose 24.

In conclusion, we have reported the stereodivergent synthesis of the first members of a new class of carbasugars by way of an efficient intramolecular SmI2-mediated aldehyde−alkene coupling using a combination of LiBr/HFIP additives. In this key step, despite the high density of functional groups, cyclobutanes with four contiguous asymmetric centers are generated with a high level of stereocontrol in reproducible yields. Beyond the

synthesis of carbaoxetanoses and analogues, this process may find application in the stereocontrolled synthesis of functionalized cyclobutanes.

EXPERIMENTAL SECTION

(3R,4S)-3,4-Bis(benzyloxy)dihydrofuran-2(3H)-one (7). To lactone 6 (12.89 mmol, 1 equiv) in dry CH_3CN (73 mL) were added BnBr (7.7 mL, 64.5 mmol, 5 equiv) and $CaSO_4$ (8.77 g, 64.5 mmol, 5 equiv). The solution was stirred for 5 min, and the flask was covered with aluminum foil. Ag₂O (5.97 g, 25.8 mmol, 2 equiv) was added in 3 portions over 5 min. The solution was stirred for 12 h, at which point a second portion of Ag₂O (5.97 g, 25.78 mmol, 2 equiv) was added. The resulting mixture was stirred for 36 h. The reaction mixture was then filtered through a pad of Celite, and the resulting filter cake was washed with CH₃CN (3×30 mL). The solution was concentrated under reduced pressure. The crude product was purified by flash chromatography (EtOAc/petroleum ether, 1:9 to 1:3) to afford the desired lactone 7 (3.04 g, 79%) as a colorless oil. Spectroscopic data are in accordance with literature data.¹⁴

(3R,4S)-3,4-Bis(benzyloxy)tetrahydrofuran-2-ol (8). DIBAL-H (1 M in hexane, 9 mL, 9.0 mmol, 1.9 equiv) was added to [a s](#page-6-0)olution of lactone 7 (1.37 g, 4.59 mmol, 1 equiv) in THF (7 mL) cooled to −78 °C. The solution was stirred at −78 °C for 2.5 h. MeOH (0.69 mL) was slowly added, and the reaction mixture was warmed up to rt. After 5 min, saturated aqueous sodium potassium tartrate (6 mL) was added. The solution was stirred overnight. The organic phase was washed with brine, dried over MgSO₄, filtered and concentrated under reduced pressure. The crude product was purified by flash chromatography (EtOAc/petroleum ether, 1:3 to 1:2) to afford the desired lactol 8 (1.18 g, 85%) as a colorless oil: TLC R_f 0.19 (silica gel, EtOAc/petroleum ether, 1:3); ¹H NMR (300 MHz, CDCl₃) δ 7.42– 7.24 (m, 10H), 5.44 (dd, J = 8.5, 4.2 Hz, 0.4H, α isomer), 5.33 (d, J = 9.9 Hz, 0.6H, β isomer), 4.66−4.60 (m, 2H), 4.56−4.47 (m, 2H), 4.22−4.05 (m, 2.6H), 4.01−3.94 (m, 1H), 3.84−3.78 (m, 0.4H, α isomer), 3.69 (d, J = 8.6 Hz, 0.4H, α isomer), 3.29 (d, J = 10.1 Hz, 0.6H, β isomer). Spectroscopic data are in accordance with literature data of its enantiomer. 21

(4S,5S,E)-Ethyl 4,5-bis(benzyloxy)-6-hydroxyhex-2-enoate ((E)-9) and (4S,5S,Z)[-Et](#page-6-0)hyl 4,5-bis(benzyloxy)-6-hydroxyhex-2 **enoate ((Z)-9).** To a solution of lactol $\frac{8}{1.85}$ g, 6.17 mmol, 1 equiv) in CH_2Cl_2 (31 mL), was added (ethoxycarbonylmethylene)triphenylphosphorane 11 (2.58 g, 7.40 mmol, 1.2 equiv) followed by benzoic acid (23 mg, 1.85 mmol, 0.03 equiv). The mixture was refluxed for 15 h. After cooling, the solution was concentrated under reduced pressure. The crude product was purified by flash chromatography (EtOAc/petroleum ether, 1:3 to 1:1) to afford the desired alcohol 9 (2.22 g, 97%, Z/E (1:2)) as a colorless oil. (E) -9: TLC R_f 0.38 (silica gel, EtOAc/petroleum ether, 1:2); $[\alpha]_{D}^{20} = +7$ (c 1.1, CHCl₃); IR (film) 3417, 2873, 1715 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.40−7.24 (m, 10H), 6.95 (dd, J = 15.9, 5.9 Hz, 1H), 6.11 $(dd, J = 15.9, 1.3 Hz, 1H), 4.72 (d, J = 11.7 Hz, 1H), 4.65 (d, J = 11.7$ Hz, 1H), 4.63 (d, J = 11.7 Hz, 1H), 4.42 (d, J = 11.7 Hz, 1H), 4.27− 4.18 (m, 3H), 3.73 (dd, J = 10.9, 3.8 Hz, 1H), 3.68−3.53 (m, 2H), 1.32 (t, $J = 7.1$ Hz, $3H$); ¹³C NMR (75 MHz, CDCl₃) δ 166.0, 144.2, 138.1, 137.7, 128.6, 128.1, 128.0, 123.8, 80.7, 78.6, 73.5, 71.9, 62.0, 60.7, 14.4; HRMS (ESI) m/z 393.166 ([M + Na]⁺, calcd. for $C_{22}H_{26}O_5$ Na 393.167).

(Z)-9. Data: TLC R_f 0.43 (silica gel, EtOAc/petroleum ether, 1:2); $[\alpha]_D^{20} = +16$ (c 1.0, CHCl₃); IR (film) 3463, 2870, 1715 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.31–7.17 (m, 10H), 6.22 (dd, J = 11.7, 9.1 Hz, 1H), 5.92 (d, $J = 11.7$ Hz, 1H), 5.27 (dd, $J = 9.0$, 4.0 Hz, 1H), 4.68 (d, $J = 11.7$ Hz, 1H), 4.58 (d, $J = 11.6$ Hz, 1H), 4.56 (d, $J = 11.7$ Hz, 1H), 4.39 (d, J = 11.7 Hz, 1H), 4.08 (q, J = 7.1 Hz, 2H), 3.75− 3.68 (m, 3H), 2.45−2.17 (br s, 1H), 1.20 (t, J = 7.1 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 166.0, 146.8, 138.4, 138.0, 128.5, 128.4, 128.10, 128.07, 127.9, 127.8, 123.2, 80.9, 75.6, 73.2, 71.9, 62.1, 60.6, 14.2; HRMS (ESI) m/z 393.167 ([M + Na]⁺, calcd. for $C_{22}H_{26}O_5Na$ 393.167).

(4S,5R,E)-Ethyl 4,5-bis(benzyloxy)-6-oxohex-2-enoate ((E)- 10) and (4S,5R,Z)-Ethyl 4,5-bis(benzyloxy)-6-oxohex-2-enoate $((Z)-10)$. A solution of DMSO $(0.60 \text{ mL}, 8.50 \text{ mmol}, 4.6 \text{ equiv})$ in $CH₂Cl₂$ (4.6 mL) was slowly added to a solution of oxalyl chloride (0.35 mL, 4.07 mmol, 2.2 equiv) in CH_2Cl_2 (8.8 mL) cooled to -78 °C. The solution was stirred for 30 min. A solution of alcohol 9 (685 mg, 1.85 mmol, 1 equiv) in CH_2Cl_2 (8.8 mL) was slowly added. The solution was stirred for 1 h. A solution of NEt_3 (1.4 mL, 10.16 mmol, 5.5 equiv) in CH_2Cl_2 (8.8 mL) was slowly added, and the solution was stirred for 1.25 h. The reaction mixture was warmed up to −20 °C. Water (58 mL) was added, and the product was extracted with CH_2Cl_2 (3×). The combined organic layer was dried over Na_2SO_4 , filtered and concentrated under reduced pressure. The crude product was purified by flash chromatography (EtOAc/petroleum ether, 1:7 to 1:2) to afford the desired aldehyde 10 (554 mg, 79%, Z/E (1:2)) as a pale yellow oil.

(E)-10. Data: TLC R_f 0.59 (silica gel, EtOAc/petroleum ether, 1:2); $[\alpha]_D^{20} = +61$ (c 1.0, CHCl₃); IR (film) 1717 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 9.66 (d, J = 1.0 Hz, 1H), 7.40–7.20 (m, 10H), 6.95 (dd, J = 15.7, 7.2 Hz, 1H), 6.09 (dd, J = 15.9, 1.0 Hz, 1H), 4.80−4.55 $(m, 4H)$, 4.35 $(m, 1H)$, 4.23 $(q, J = 7.2 \text{ Hz}, 2H)$, 3.83 $(dd, J = 3.9, 1.1$ Hz, 1H), 1.32 (t, J = 7.1 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 202.0, 165.7, 143.1, 137.0, 136.7, 128.7, 128.6, 128.44, 128.41, 128.22, 128.16, 124.5, 84.0, 77.9, 73.7, 72.0, 60.8, 14.4; HRMS (ESI) m/z 391.151 ($[M + Na]^+$, calcd. for C₂₂H₂₄O₅Na 391.152).

(Z)-10. Data: TLC R_f 0.41 (silica gel, EtOAc/petroleum ether, 1:2); $[\alpha]_{\text{D}}^{20}$ = +89 (c 1.0, CHCl₃); IR (film) 1732, 1714 cm⁻¹; ¹H NMR $(300 \text{ MHz}, \text{CDCl}_3)$ δ 9.60 $(d, J = 1.0 \text{ Hz}, 1H)$, 7.35–7.16 $(m, 10H)$, 6.31 (dd, $J = 11.7$, 8.1 Hz, 1H), 5.86 (dd, $J = 11.7$, 1.5 Hz, 1H), 5.41 (ddd, J = 8.1, 3.6, 1.4 Hz, 1H), 4.72 (d, J = 12.1 Hz, 1H), 4.57−4.48 $(m, 2H)$, 4.34 (d, J = 11.9 Hz, 1H), 4.07 (q, J = 7.2 Hz, 2H), 3.99 (dd, $J = 3.6, 1.1$ Hz, 1H), 1.20 (t, $J = 7.1$ Hz, 3H); ¹³C NMR (75 MHz, CDCl3) δ 201.5, 165.6, 146.8, 137.4, 137.1, 128.53, 128.48, 128.4, 128.23, 128.21, 128.0, 122.8, 85.3, 75.1, 73.8, 72.2, 60.7, 14.3; HRMS (ESI) m/z 391.151 ([M + Na]⁺, calcd. for C₂₂H₂₄O₅Na 391.152).

Ethyl 2-((1R,2S,3S,4R)-2,3-bis(benzyloxy)-4-hydroxycyclo**butyl)acetate (12).** SmI_2 (0.1 M in THF, 12.5 mL, 1.25 mmol, 4 equiv) was added to LiBr (325 mg, 3.74 mmol, 12 equiv) in a flask covered with aluminum foil. The solution was stirred for 20 min and then cooled to −78 °C. HFIP (0.26 mL, 2.50 mmol, 8 equiv) followed by a solution of aldehyde 10 (115 mg, 0.31 mmol, 1 equiv) in degassed THF (10.6 mL) were added. The solution was stirred for 5 min, and HCl 1 N (15.5 mL) was added. The solution was then stirred at rt for 30 min. The product was extracted with CH_2Cl_2 (3×). The combined organic layer was washed with saturated aqueous $NaHSO₃$, dried over MgSO4, filtered and concentrated under reduced pressure. The crude product was purified by flash chromatography (EtOAc/petroleum ether, 1:5 to 1:1) to afford the desired cyclobutane 12 (78 mg, 67%) as a yellow oil: TLC R_f 0.34 (silica gel, EtOAc/petroleum ether, 1:2); $[\alpha]_{D}^{20} = -32$ (c 1.0, CHCl₃); IR (film) 3444, 1731 cm⁻¹; ¹H NMR $(300 \text{ MHz}, \text{CDCl}_3)$ δ 7.34–7.17 (m, 10H), 4.62 (d, J = 11.7 Hz, 1H), 4.57−4.47 (m, 2H), 4.42 (d, J = 11.9 Hz, 1H), 4.07 (q, J = 7.2 Hz, 2H), 3.81 (t, $J = 5.8$ Hz, 1H), 3.51 (t, $J = 6.6$ Hz, 1H), 3.32 (m, 1H), 3.17 (br s, 1H), 2.65 (dd, J = 16.7, 4.7 Hz, 1H), 2.65 (dd, J = 16.7, 10.7) Hz, 1H), 1.93 (m, 1H), 1.19 (t, J = 7.2 Hz, 3H); ¹³C NMR (75 MHz, CDCl3) δ 173.5, 138.2, 138.1, 128.6, 128.5, 128.0, 127.9, 127.8, 85.1, 76.9, 72.3, 71.63, 71.56, 61.1, 40.2, 36.7, 14.3; HRMS (ESI) m/z 393.167 ($[M + Na]^+$, calcd. for $C_{22}H_{26}O_5Na$ 393.167).

(1R,2S,3S,4R)-2,3-Bis(benzyloxy)-4-(2-hydroxyethyl)cyclobutanol (14). LAH (13 mg, 0.33 mmol, 1.5 equiv) was added to a solution of ester 12 (82 mg, 0.22 mmol, 1 equiv) in THF (1.2 mL) cooled at 0 °C. The solution was stirred at rt for 2.5 h. After cooling at 0 °C, H_2O (0.01 mL) followed by aqueous 10% NaOH (0.02 mL) and H2O (0.03 mL) were added. The solution was filtered and concentrated under reduced pressure. The crude product was purified by flash chromatography (EtOAc/petroleum ether, 1:1 to 4:1) to afford the desired diol 14 (50 mg, 68%) as a white solid: TLC R_f 0.16 (silica gel, EtOAc/petroleum ether, 2:1); $[\alpha]_D^{20} = -9$ (c 1.0, CHCl₃); IR (film) 3379 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.41-7.25 (m, 10H), 4.67−4.57 (m, 3H), 4.53 (d, J = 11.7 Hz, 1H), 3.78 (t, J = 5.9

Hz, 1H), 3.71 (m, 1H), 3.61 (m, 1H), 3.42 (t, $J = 6.5$ Hz, 1H), 3.36 $(m, 1H)$, 2.96–2.75 (br s, 2H), 1.79 $(m, 1H)$, 1.73–1.59 $(m, 2H)$; ¹³C NMR (75 MHz, CDCl3) δ 138.3, 138.0, 128.6, 128.5, 128.03, 127.95, 127.9, 85.8, 76.9, 72.4, 71.71, 71.66, 62.0, 42.7, 34.6; HRMS (ESI) m/z 351.155 ($[M + Na]^+$, calcd. for $C_{20}H_{24}O_4$ Na 351.157).

General Method A for Debenzylation. Pd/C 10% (10% weight) and $HCO₂H$ (2 drops) were added to a solution of cyclobutane derivatives in EtOH (0.083 M). The solution was placed under H_2 atmosphere and stirred until disappearance of the starting material (16 h). The solution was filtered through Celite and concentrated under reduced pressure. The crude product was purified by flash chromatography.

(1R,2r,3S,4s)-4-(2-Hydroxyethyl)cyclobutane-1,2,3-triol (15). According to general method A, 14 (250 mg, 0.76 mmol) afforded the tetrol 15 (112 mg, quant.) as a colorless oil: TLC R_f 0.19 (silica gel MeOH/CH2Cl2, 2:8); IR (neat) 3274, 1050, 1024 cm⁻¹; ¹H NMR $(300 \text{ MHz}, \text{ MeOD}) \delta 3.65$ (t, J = 6.9 Hz, 2H), 3.55 (t, J = 6.3 Hz, 1H), 3.17 (dd, J = 7.8, 6.2 Hz, 2H), 1.79 (q, J = 7.1 Hz, 2H), 1.47 (m, 1H); 13C NMR (75 MHz, MeOD) δ 81.7, 73.8, 61.5, 44.0, 36.0; HRMS (ESI) m/z 171.065 ([M + Na]⁺, calcd. for $C_6H_{12}O_4Na$ 171.063).

(1R,2S,3S,4R)-2,3-Bis(benzyloxy)-4-vinylcyclobutanol (16). To a solution of diol 14 (200 mg, 0.61 mmol, 1 equiv) in THF (20 mL) were added 2-nitrophenyl selenocyanate (304 mg, 1.34 mmol, 2.2 equiv) in one portion followed by tributylphosphine (0.33 mL, 1.34 mmol, 2.2 equiv) dropwise. The solution was stirred at rt for 35 min. Water was added, and the product was extracted with $Et₂O (3x)$. The combined organic layer was dried over MgSO4, filtered and concentrated under reduced pressure. A solution of the crude selenyl derivative in CH_2Cl_2 (8 mL) was added to a solution of Davis oxaziridine^{20b} (191 mg, 0.73 mmol, 1.2 equiv) in CH_2Cl_2 (10 mL) cooled to 0 °C. The solution was stirred at 0 °C for 40 min. Saturated aqueous Na_2CO_3 (6 mL) was added, and the solution was stirred at rt for 1 h. The product was extracted with CH_2Cl_2 (3×). The combined organic layer was washed with saturated aqueous $NaHSO₃$, dried over MgSO4, filtered and concentrated under reduced pressure. The crude product was purified by flash chromatography (EtOAc/petroleum ether, 1:7 to 1:3) to afford the desired vinylcyclobutanol 16 (160 mg, 84%) as a yellow oil: TLC R_f 0.49 (silica gel, EtOAc/petroleum ether, 1:2); $[\alpha]_D^{20} = +14$ (c 1.0, CHCl₃); IR (film) 3392, 2871, 1087 cm⁻¹;
¹H NMP (200 MHz, CDCl) δ 733–716 (m, 10H) 5.80 (ddd 1 – ¹H NMR (300 MHz, CDCl₃) δ 7.33–7.16 (m, 10H), 5.80 (ddd, J = 17.3, 10.1, 7.4 Hz, 1H), 5.09 (d, $J = 17.0$ Hz, 1H), 5.01 (d, $J = 10.2$ Hz, 1H), 4.59 (d, J = 11.9 Hz, 1H), 4.52 (d, J = 12.0 Hz, 1H), 4.51 (d, J = 12.0 Hz, 1H), 4.44 (d, $J = 11.9$ Hz, 1H), 3.73 (t, $J = 5.9$ Hz, 1H), 3.50 (m, 1H), 3.42 (m, 1H), 2.25 (q, J = 7.7 Hz, 1H), 2.19−2.10 (br s, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 138.2, 138.0, 137.0, 128.6, 128.5, 128.1, 128.0, 127.9, 127.8, 116.2, 85.4, 76.9, 71.83, 71.81, 71.4, 48.3; HRMS (ESI) m/z 333.145 ([M + Na]⁺, calcd. for $C_{20}H_{22}O_3Na$

333.146).
((1R,2R,3S,4S)-2,3-Bis(benzyloxy)-4-vinylcyclobutoxy)(tertbutyl)dimethylsilane (17). To a solution of 16 (99 mg, 0.32 mmol, 1 equiv) in CH_2Cl_2 (0.61 mL) cooled to 0 °C were added TBSCl (72 mg, 0.48 mmol, 1.5 equiv), DMAP (20 mg, 0.16 mmol, 0.5 equiv) and NEt_3 (0.09 mL, 0.64 mmol, 2 equiv). The solution was stirred at rt for 14 h. Water was added, and the product was extracted with CH_2Cl_2 (3×). The combined organic layer was dried over Na_2SO_4 , filtered and concentrated under reduced pressure. The crude product was purified by flash chromatography (EtOAc/petroleum ether, 3:97) to afford the desired cyclobutane 17 (105 mg, 77%) as a yellow oil: TLC R_f 0.42 (silica gel, EtOAc/petroleum ether, 1:19); $[\alpha]_D^{20} = +6.5$ (c 1.0, CHCl₃); IR (film) 2928, 2857, 1096 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.42−7.22 (m, 10H), 5.87 (ddd, J = 17.2, 10.1, 8.3 Hz, 1H), 5.14 (d, J = 17.2 Hz, 1H), 5.07 (d, J = 10.1 Hz, 1H), 4.65 (d, J = 11.7 Hz, 1H), 4.59 (d, J = 11.7 Hz, 1H), 4.57 (d, J = 11.9 Hz, 1H), 4.48 (d, $J = 11.7$ Hz, 1H), 3.83 (t, $J = 5.9$ Hz, 1H), 3.62 (t, $J = 6.5$ Hz, 1H), 3.49 (t, J = 6.3 Hz, 1H), 2.33 (q, J = 7.6 Hz, 1H), 0.90 (s, 9H), 0.07 (s, 3H), 0.05 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 138.31, 138.26, 137.8, 128.5, 127.9, 127.7, 116.2, 85.5, 77.4, 72.4, 71.6, 71.3, 49.0, 25.9, 18.0, -4.0, -4.6. HRMS (ESI) m/z 447.236 ([M + Na]⁺, calcd. for $C_{26}H_{36}O_3SiNa$ 447.233).

((1S,2S,3R,4R)-2,3-Bis(benzyloxy)-4-((tert-butyldimethylsilyl)oxy)cyclobutyl)methanol (18). To a solution of 17 (99 mg, 0.23 mmol, 1 equiv) in a mixture of THF/ H_2O (4:1, 0.7 mL) were added $OsO₄$ (2.5 wt % in tBuOH, 0.09 mL, 0.009 mmol, 0.04 equiv) and NaIO_4 (94 mg, 0.44 mmol, 1.9 equiv). The solution was stirred at rt for 5 h. Saturated aqueous $\text{Na}_2\text{S}_2\text{O}_3$ was added, and the product was extracted with EtOAc (3×). The combined organic layer was dried over Na2SO4, filtered and concentrated under reduced pressure. The residue was dissolved in MeOH, and the solution was cooled to 0 °C. $NaBH₄$ (11 mg, 0.29 mmol, 1.2 equiv) was added, and the solution was stirred at rt for 15 min. Acetone (0.5 mL) was added. The solution was stirred 5 min and then concentrated under reduced pressure. The residue was dissolved in EtOAc and H_2O , and the product was extracted with EtOAc (4×). The combined organic layer was dried over $Na₂SO₄$, filtered and concentrated under reduced pressure. The crude product was purified by flash chromatography (EtOAc/ petroleum ether, 3:97) to afford the desired alcohol 18 (82 mg, 82%) as a colorless oil: TLC R_f 0.18 (silica gel, EtOAc/petroleum ether, 1:5); $[\alpha]_D^{20}$ = +0.1 (c 1.0, CHCl₃); IR (film) 3454, 2929, 2857 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.38−7.25 (m, 10H), 4.64 (d, J $= 11.6$ Hz, 1H), 4.58 (d, J = 12.7 Hz, 1H), 4.56 (s, 2H), 3.84 (t, J = 5.8 Hz, 1H), 3.76 (dd, J = 11.1, 4.5 Hz, 1H), 3.72−3.64 (m, 2H), 3.50 (m, 1H), 1.91 (m, 1H), 0.90 (s, 9H), 0.10 (s, 3H), 0.08 (s, 3H); 13C NMR (75 MHz, CDCl3) δ 138.4, 138.3, 128.6, 128.5, 127.94, 127.87, 127.8, 127.7, 85.9, 74.8, 71.7, 71.6, 68.2, 62.0, 47.0, 25.9, 18.1, −4.4, −4.6; HRMS (ESI) m/z 451.230 ([M + Na]⁺, calcd. for $C_{25}H_{36}O_4SiNa$ 451.228).

(1R,2S,3S,4R)-2,3-Bis(benzyloxy)-4-(hydroxymethyl)cyclo**butanol (19).** TBAF (1 M in THF, 3.1 mL, 3.1 mmol, 5 equiv) was added to a solution of 18 (264 mg, 0.62 mmol, 1 equiv) in THF (1 mL). The solution was stirred at rt for 4 h. Saturated aqueous NH_4Cl was added, and the product was extracted with $EtOAc$ (3 \times). The combined organic layer was dried over $Na₂SO₄$, filtered and concentrated under reduced pressure. The crude product was purified by flash chromatography (MeOH/CH₂Cl₂, 3:97 to 5:95) to afford the desired diol 19 (197 mg, quant.) as a colorless oil: TLC R_f 0.49 (silica gel, MeOH/CH₂Cl₂, 10:90); $[\alpha]_D^{20} = +21$ (c 1.0, MeOH); IR (neat) 3369, 2930, 2871 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.40–7.35 (m, 10H), 4.65 (d, J = 11.9 Hz, 1H), 4.58 (d, J = 11.9 Hz, 1H), 4.56 (d, J = 11.9 Hz, 1H), 4.52 (d, J = 11.7 Hz, 1H), 3.80 (t, J = 5.7 Hz, 1H), 3.73 $(dd, J = 11.1, 5.0 Hz, 1H), 3.65-3.55 (m, 2H), 3.45 (m, 1H), 2.68-$ 2.56 (br s, 1H), 2.18−2.07 (br s, 1H), 1.89 (m, 1H); 13C NMR (100 MHz, CDCl₃) δ 138.3, 138.2, 128.63, 128.58, 128.1, 128.01, 127.98, 127.94, 85.8, 74.2, 71.79, 71.77, 69.3, 62.4, 46.5; HRMS (ESI) m/z 337.143 ($[M + Na]^+$, calcd. for $C_{19}H_{22}O_4Na$ 337.141).

(1R,2r,3S,4s)-4-(Hydroxymethyl)cyclobutane-1,2,3-triol (5). According to general method A, 19 (80 mg, 0.25 mmol) afforded the tetrol 5 (33 mg, quant.) as a colorless oil: TLC R_f 0.28 (silica gel, $CH_3CN/H_2O/NH_4OH$, 5:1:1); IR (film) 3237, 1068 cm⁻¹; ¹H NMR (300 MHz, D_2O) δ 3.80−3.71 (m, 3H), 3.50−3.42 (m, 2H), 1.76 (m, 1H); ¹³C NMR (75 MHz, D₂O) δ 79.1, 68.1, 60.2, 46.2; HRMS (ESI) m/z 157.048 ([M + Na]⁺, calcd. for C₅H₁₀O₄Na 157.047).

((1R,2R,3S,4S)-2,3-Bis(benzyloxy)-4-((benzyloxy)methyl) cyclobutoxy)(tert-butyl)dimethylsilane (20). NaH (17 mg, 0.41 mmol, 1.3 equiv) was added to a solution of 18 (136 mg, 0.32 mmol, 1 equiv) in THF (1.7 mL) cooled at 0 °C. The solution was stirred at rt for 30 min. BnBr (45 μ L, 0.38 mmol, 1.2 equiv) was then added. The solution was stirred at rt for 18 h. Water was added, and the product was extracted with CH_2Cl_2 (3×). The combined organic layer was dried over MgSO₄, filtered and concentrated under reduced pressure. The crude product was purified by flash chromatography (EtOAc/ petroleum ether, 3:97 to 15:85) to afford the desired cyclobutane 20 (80 mg, 49%) as a colorless oil: TLC R_f 0.51 (silica gel, EtOAc/ petroleum ether, 1:9); IR (film) 2929, 2856, 1096 cm⁻¹; $[\alpha]_D^{20} = +2$ (c 1.0, CHCl₃); ¹H NMR (300 MHz, CDCl₃) δ 7.40–7.23 (m, 15H), 4.64 (d, $J = 11.7$ Hz, 1H), 4.58 (d, $J = 11.7$ Hz, 1H), 4.57 (d, $J = 12.1$ Hz, 1H), 4.54–4.46 (m, 3H), 3.84 (t, J = 5.7 Hz, 1H), 3.73 (m, 1H), 3.60−3.47 (m, 3H), 1.97 (m, 1H), 0.89 (s, 9H), 0.08 (s, 3H), 0.05 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 138.7, 138.5, 138.4, 128.5, 127.89, 127.86, 127.81, 127.7, 86.0, 75.1, 73.2, 71.6, 71.5, 68.9, 68.4,

45.3, 25.9, 18.1, −4.4, −4.6; HRMS (ESI) m/z 541.274 ([M + Na]⁺ , calcd. for $C_{32}H_{42}O_4SiNa$ 541.274).

(1R,2S,3S,4R)-2,3-Bis(benzyloxy)-4-((benzyloxy)methyl) cyclobutanol (21). TBAF $(1 \text{ M in THE}, 1.1 \text{ mL}, 1.10 \text{ mmol}, 5$ equiv) was added to a solution of 20 (113 mg, 0.22 mmol, 1 equiv) in THF (0.4 mL). The solution was stirred at rt for 3 h. Saturated aqueous NH4Cl was added, and the product was extracted with EtOAc (3 \times). The combined organic layer was dried over MgSO₄, filtered and concentrated under reduced pressure. The crude product was purified by flash chromatography (EtOAc/petroleum ether, 3:97 to 15:85) to afford the desired cyclobutanol 21 (85 mg, 97%) as a colorless oil: TLC R_f 0.29 (silica gel, EtOAc/petroleum ether, 1:2); IR (film) 3406, 2861, 1051 cm⁻¹; $\left[\alpha\right]_D{}^{20} = -3$ (c 1.0, CHCl₃); ¹H NMR (300 MHz, CDCl₃) δ 7.42−7.25 (m, 15H), 4.67 (d, J = 11.9 Hz, 1H), 4.61 (d, J = 11.9 Hz, 1H), 4.56 (s, 2H), 4.52 (s, 2H), 3.83 (t, J = 5.9 Hz, 1H), 3.65 $(m, 1H)$, 3.61–3.47 $(m, 3H)$, 2.41–2.26 (br s, 1H), 2.00 $(m, 1H)$; ¹³C NMR (75 MHz, CDCl₃) δ 138.3, 128.55, 128.50, 128.46, 128.1, 127.9, 127.7, 85.7, 74.6, 73.2, 71.7, 71.6, 69.5, 69.3, 44.8; HRMS (ESI) m/z 427.191 ($[M + Na]^+$, calcd. for $C_{26}H_{28}O_4$ Na 427.188).

(2R,3S,4S)-2,3-Bis(benzyloxy)-4-((benzyloxy)methyl)cyclo**butanone (22).** Dess–Martin periodinane (0.3 M in CH_2Cl_2 , 0.5 mL, 0.15 mmol, 1.6 equiv) was added to a solution of alcohol 21 (37.5 mg, 0.093 mmol, 1 equiv) in CH_2Cl_2 (0.35 mL) cooled at 0 °C. The solution was stirred at rt for 1 h. Saturated aqueous $Na₂S₂O₃$ was added, and the product was extracted with CH_2Cl_2 (3×). The combined organic layer was washed with saturated aqueous $NAHCO₃$, dried over MgSO₄, filtered and concentrated under reduced pressure. The crude product was purified by flash chromatography (EtOAc/ petroleum ether, 1:7 to 1:5) to afford the desired cyclobutanone 22 (34.5 mg, 92%) as a colorless oil: TLC R_f 0.62 (silica gel, EtOAc/ petroleum ether, 1:2); IR (film) 2862, 1786, 1116 cm⁻¹; $[\alpha]_D^2$ ²⁰ = +33 (c 1.0, CHCl₃); ¹H NMR (300 MHz, C₆D₆) δ 7.31–7.01 (m, 15H), 4.69 (d, $J = 11.9$ Hz, 1H), 4.58 (m, 1H), 4.54 (d, $J = 11.6$ Hz, 1H), 4.49 (d, $J = 12.2$ Hz, 1H), 4.44 (d, $J = 12.2$ Hz, 1H), 4.31 (m, 1H), 4.18 (s, 2H), 3.35 (d, $J = 2.0$ Hz, 1H), 3.33 (d, $J = 0.7$ Hz, 1H), 2.84 (m, 1H); ¹³C NMR (75 MHz, C_6D_6) δ 203.2, 138.6, 138.5, 138.0, 128.63, 128.61, 128.2, 127.9, 127.8, 91.3, 74.4, 73.2, 72.4, 72.2, 65.8, 58.2; HRMS (ESI) m/z 425.174 ([M + Na]⁺, calcd. for $C_{26}H_{26}O_4$ Na 425.172).

(1S,2S,3S,4R)-2,3-Bis(benzyloxy)-4-((benzyloxy)methyl) cyclobutanol (23). L-Selectride (1 M in THF, 0.18 mL, 0.18 mmol, 1.1 equiv) was added to a solution of ketone 22 (66 mg, 0.16 mmol, 1 equiv) in THF (0.66 mL) cooled to −78 °C. The solution was stirred for 1 h. Water (0.01 mL) was added, and the solution was warmed up to 0 °C. Aqueous 35% H_2O_2 (0.02 mL) was added, and the solution was diluted in EtOAc. The organic layer was washed with saturated aqueous $Na₂SO₃$, saturated aqueous $NaHCO₃$ and brine. The organic layer was dried over MgSO₄, filtered and concentrated under reduced pressure. The crude product was purified by flash chromatography (EtOAc/petroleum ether, 1:5 to 1:2) to afford the desired cyclobutanol 23 (44 mg, 67%) as a colorless oil: TLC R_f 0.34 (silica gel, EtOAc/petroleum ether, 1:2); IR (film) 3455, 2863, 1100 cm⁻¹; $[\alpha]_D^{20} = +9$ (c 1.0, CHCl₃); ¹H NMR (300 MHz, CDCl₃) δ 7.39– 7.24 (m, 15H), 4.57 (s, 2H), 4.54−4.46 (m, 4H), 4.31 (m, 1H), 4.04 $(dd, J = 8.2, 6.2 \text{ Hz}, 1H), 3.87 \text{ (m, 1H)}, 3.73 \text{ (dd, } J = 9.7, 7.4 \text{ Hz}, 1H),$ 3.63 (dd, $J = 9.7$, 5.5 Hz, 1H), 2.40 (d, $J = 4.1$ Hz, 1H), 2.19 (m, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 138.4, 138.3, 137.8, 128.7, 128.5, 128.21, 128.18, 127.9, 127.85, 127.80, 127.76, 81.3, 78.1, 73.3, 72.0, 71.9, 67.4, 64.7, 41.6; HRMS (ESI) m/z 427.187 ([M + Na]⁺, calcd. for $C_{26}H_{28}O_4$ Na 427.188).

(1S,2R,3S,4S)-4-(Hydroxymethyl)cyclobutane-1,2,3-triol (24). General method A was used, and 23 (44 mg, 0.11 mmol) afforded the tetrol 24 (14.5 mg, quant.) as a white solid: TLC R_f 0.26 (silica gel, $\rm CH_3CN/H_2O/NH_4OH$, 5:1:1); IR (neat) 3207, 1051 cm⁻¹; $\rm [\alpha]_{\rm D}$ $20 =$ -42 (c 0.5, H₂O); ¹H NMR (300 MHz, D₂O) δ 4.32 (m, 1H), 3.99– 3.87 (m, 2H), 3.87−3.71 (m, 2H), 2.04 (m, 1H); 13C NMR (75 MHz, D2O) δ 74.8, 72.0, 64.8, 58.0, 43.0; HRMS (ESI) m/z 157.047 ([M + Na]⁺, calcd. for $C_5H_{10}O_4$ Na 157.047).

■ ASSOCIATED CONTENT

S Supporting Information

 1 H NMR and 13 C NMR for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

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■ REFERENCES

(1) (a) Sears, P.; Wong, C.-H. Angew. Chem., Int. Ed. 1999, 38, 2300−2324. (b) Compain, P.; Martin, O. R. Bioorg. Med. Chem. 2001, 9, 3077−3092. (c) Ernst, B.; Magnani, J. L. Nat. Rev. Drug Discovery 2009, 8, 661−677. (d) Koester, D. C.; Holkenbrink, A.; Werz, D. B. Synthesis 2010, 3217−3242. (e) Chapleur, Y. Carbohydrate Mimics, Concepts and Methods; Wiley-VCH: Weinheim, 1998.

(2) McCasland, G. E.; Furuta, S.; Durham, L. J. J. Org. Chem. 1966, 31, 1516−1521.

(3) (a) Suami, T.; Ogawa, S. Adv. Carbohydr. Chem. Biochem. 1990, 48, 21−90. (b) Ogawa, S. In Carbohydrate Mimics, Concepts and Methods; Wiley-VCH: Weinheim, 1998; pp 87−106. (c) Arjona, O.; Gomez, A. M.; Cristobal Lopez, J.; Plumet, J. Chem. Rev. 2007, 107, 1919−2036.

(4) (a) Ogawa, S.; Kanto, M.; Suzuki, Y. Mini-Rev. Med. Chem. 2007, 7, 679−691. (b) Chen, X.; Fan, Y.; Zheng, Y.; Shen, Y. Chem. Rev. 2003, 103, 1955−1978. (c) Mahmud, T. Nat. Prod. Rep. 2003, 20, 137−166. (d) Crimmins, M. T. Tetrahedron 1998, 54, 9229−9272. (e) Berecibar, A.; Grandjean, C.; Siriwardena, A. Chem. Rev. 1999, 99, 779−844. (f) Schiffelers, R.; Storm, G.; Bakker-Woudenberg, I. J. Antimicrob. Chemother. 2001, 48, 333−344. (g) Mingeot-Leclercq, M.- P.; Glupczynski, Y.; Tulkens, P. M. Antimicrob. Agents Chemother. 1999, 43, 727−737.

(5) Chiasson, J. L.; Josse, R. G.; Gomis, R.; Hanefeld, M.; Karasik, A.; Laakso, M. Lancet 2002, 359, 2072−2077.

(6) Chen, X.; Zheng, Y.; Shen, Y. Curr. Med. Chem. 2006, 13, 109− 116.

(7) (a) Ramana, C. V.; Chaudhuri, S. R.; Gurjar, M. K. Synthesis 2007, 523−528. (b) Rassu, G.; Auzzas, L.; Pinna, L.; Zambrano, V.; Zanardi, F.; Battistini, L.; Gaetani, E.; Curti, C.; Casiraghi, G. J. Org. Chem. 2003, 68, 5881−5885. (c) Sisu, E.; Sollogoub, M.; Mallet, J.-M.; Sinaÿ, P. Tetrahedron 2002, 58, 10189−10196. (d) Marco-Contelles, J.; De Opazo, E. J. Carbohydr. Chem. 2002, 21, 201−218. (e) Marco-Contelles, J.; De Opazo, E. Tetrahedron Lett. 1999, 40, 4445−4448.

(8) (a) Paquette, L. A.; Moura-Letts, G.; Wang, G. P. J. Org. Chem. 2009, 74, 2099−2107. (b) Paquette, L. A.; Zhang, Y. J. Org. Chem. 2006, 71, 4353−4363. (c) Mehta, G.; Pallavi, K. Chem. Commun. 2002, 2828−2829. (d) Wang, W.; Zhang, Y.; Zhou, H.; Blériot, Y.; Sinaÿ, P. Eur. J. Org. Chem. 2001, 1053-1059. (e) Wang, W.; Zhang, Y.; Sollogoub, M.; Sinaÿ, P. Angew. Chem., Int. Ed. Engl. 2000, 39, 2466−2467.

(9) Mehta, G.; Pallavi, K.; Katukojvala, S. Tetrahedron Lett. 2009, 50, 4519−4522.

(10) For recent examples, see: (a) Compain, P.; Decroocq, C.; Iehl, J.; Holler, M.; Hazelard, D.; Mena Barragán, T.; Ortiz Mellet, C.; Nierengarten, J.-F. Angew. Chem., Int. Ed. 2010, 49, 5753−5756. (b) Chagnault, V.; Compain, P.; Lewinski, K.; Ikeda, K.; Asano, N.; Martin, O. R. J. Org. Chem. 2009, 74, 3179−3182.

The Journal of Organic Chemistry Note

(11) (a) For examples of sugar oxetanoses, see: Nishiyama, S.; Ohgiya, T.; Yamamura, S.; Kato, K.; Nagai, M.; Takita, T. Tetrahedron Lett. 1990, 31, 705−708. (b) For examples of cyclobutyl nucleosides, see: Norbeck, D. W.; Kern, E.; Hayashi, S.; Rosenbrook, W; Sham, H.; Herrin, T.; Plattner, J. J.; Erickson, J.; Clement, J.; Swanson, R.; Shipkowitz, N.; Hardy, D.; Marsh, K.; Arnett, G.; Shannon, W.; Broder, S.; Mitsuya, H. J. Med. Chem. 1990, 33, 1281−1285.

 (12) (a) For a review on SmI₂-mediated carbonyl-alkene coupling for the synthesis of small carbocyclic rings, see: Harb, H. Y.; Procter, D. J. Synlett 2012, 6−20. (b) Williams, D. B. G.; Blann, K. Eur. J. Org. Chem. 2004, 3286−3291. (c) Williams, D. B. G.; Caddy, J.; Blann, K.; Grove, J. J. C.; Holzapfel, C. W. ́ Synthesis 2009, 2009−2014.

(13) Schö ning, K.-U; Scholz, P.; Wu, X.; Guntha, S.; Delgado, G.; Krishnamurthy, R.; Eschenmoser, A. Helv. Chim. Acta 2002, 85, 4111− 4153.

(14) Spectroscopic and analytical data of lactone 7 are in good agreement with literature values, see: Pabba, J.; Rempel, B. P.; Withers, S. G.; Vasella, A. Helv. Chim. Acta 2006, 89, 635−666.

(15) Vu, N. Q.; Chai, C. L. L.; Lim, K. P.; Chia, S. C.; Chen, A. Tetrahedron 2007, 63, 7053−7058.

(16) HMPA was used as an additive to generate an activated HMPA-SmI2 complex, see: (a) Weinges, K.; Schmidbauer, S. B.; Schick, H. Chem. Ber. 1994, 127, 1305−1309. (b) Molander, G. A.; McKie, J. A. J. Org. Chem. 1992, 57, 3132−3139. (c) Sadasivam, D. V.; Antharjanam, P. K. S.; Prasad, E.; Flowers, R. A., II J. Am. Chem. Soc. 2008, 130, 7228−7229.

(17) Johnston, D.; McCusker, C. M.; Procter, D. J. Tetrahedron Lett. 1999, 40, 4913−4916.

(18) (a) Chopade, P. R.; Prasad, E.; Flowers, R. A. J. Am. Chem. Soc. 2004, 126, 44−45. (b) Edmonds, D. J.; Muir, K. W.; Procter, D. J. J. Org. Chem. 2003, 68, 3190−3198. (c) Nicolaou, K. C.; Li, A.; Edmonds, D. J.; Tria, G. S.; Ellery, S. P. J. Am. Chem. Soc. 2009, 131, 16905−16918.

(19) Gilles, P.; Py, S. Org. Lett. 2012, 14, 1042−1045.

(20) (a) Grieco, P. A.; Gilman, S.; Nishizawa, M. J. Org. Chem. 1976, 41, 1485−1486. (b) For the preparation of Davis oxaziridine, see: Vishwakarma, L. C.; Stringer, O. D.; Davis, F. A. Org. Synth. 1988, 66, 203−210.

(21) Buchotte, M.; Muzard, M.; Plantier-Royon, R. Eur. J. Org. Chem. 2008, 3529−3534.